

ambient conditions. In this frame, our group has recently provided several examples of carbon materials syntheses but fewer ones in the direction of inorganics, the latter mainly focusing on the action of fuming nitric acid HNO_3 on organometallic precursors, such as metallocenes and metallocene dichlorides. Here special attention is given to the investigation of additional chemical options towards the hypergolic synthesis of inorganic materials with unusual shapes. The proposed alternatives include, but not limited to, (1) the hypergolic ignition of titanium (IV) dimethylamide $\text{Ti}[\text{N}(\text{CH}_3)_2]_4$ by fuming nitric acid HNO_3 towards the formation of titania TiO_2 nanoparticles with twin morphology, (2) the hypergolic ignition of molten iron (III) chloride hexahydrate $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$ by sodium hydride NaH towards the formation of thin maghemite $\gamma\text{-Fe}_2\text{O}_3$ nanosheets. Worth noting, tin (IV) dimethylamide $\text{Sn}[\text{N}(\text{CH}_3)_2]_4$ reacts similarly to titanium (IV) dimethylamide $\text{Ti}[\text{N}(\text{CH}_3)_2]_4$ with fuming nitric acid HNO_3 to correspondingly afford conductive stannic oxide SnO_2 , whereas cobalt (II) chloride hexahydrate $\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$ and copper (II) nitrate hexahydrate $\text{Cu}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ react similarly to iron (III) chloride hexahydrate $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$ with sodium hydride NaH to correspondingly afford metallic cobalt Co and copper (II) oxide CuO , respectively. Therefore, metal (IV) dimethylamides and low melting point hydrated transition metal salts (chlorides or nitrates) appear as a new and general class of precursors towards the hypergolic synthesis of inorganic materials with interesting morphologies.

Conflict of interest

The author declared that he has no conflicts of interest to this work.

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